Kinetic Analysis Reveals a Shift In Inter-subunit Communication Within F₄₂₀H₂:NADP⁺ Oxidoreductase



Abstract

 $F_{420}H_2$:NADP⁺ Oxidoreductase (Fno) catalyzes the reversible reduction of NADP⁺ to NADPH, using reduced F_{420} cofactor as the hydride donor. NADPH and the F_{420} cofactor, are linked to several metabolic pathways including glycolysis and methanogenesis within methanogenic and sulfate-reducing archaea. Previous steady-state kinetic data conducted on *wt*Fno displayed a Lineweaver-Burk plot, which was downward and concave in curvature, indicative of negative cooperativity. The pre steady-state kinetic studies displayed biphasic kinetics with an initial burst phase followed by a subsequent slow phase. The amplitude of the burst phase revealed 50% F_{420} cofactor reduction, indicating half-site reactivity. These studies suggest that Fno regulates NADPH production within the cell. The goal of this project is to determine which amino acids are involved in communication at the subunit interface. We have identified four amino acids which are Arg186, Thr192, Ser190, and His133. We have designed a library of Fno variants to test our hypothesis. The generated Fno variants (R186Q, R186K, R186I, T192V, T192A, H133A, H133N and S190A) were then characterized using fluorescence binding, steady-state and pre steady-state kinetic experiments. Our studies revealed that unlike wtFno, the following Fno variants, R186Q, R186K, R186I, T192A, and S190A displayed no cooperativity. This suggests that R186, T192 and S190 are involved in subunit communication. The results are reported here.

Introduction

 F_{420} -dependent enzymes are important in several organisms, which play vital roles in methane production, NADPH regulation, nucleic acid biosynthesis, folate biosynthesis and carbon cycling. Until our recent publications on F₄₂₀H₂:NADP⁺ Oxidoreductase (Fno) and F₄₂₀-dependent glucose-6phosphate dehydrogenase (FGD), these enzymes, in general, had not been subjected to rigorous enzymological investigation. Our work has provided valuable new mechanistic and functional insights into the enzymes that use this unique cofactor. Our previous kinetic studies on Fno, which is the focus of this proposal, indicated half-site reactivity in only one of the active sites in a functional dimer. These data suggest that Fno participates in negative cooperativity kinetics and that this enzyme regulates NADPH production methanogenic organisms. Based upon our kinetic studies, we have proposed a chemically plausible mechanism and have identified several key amino acids that potentially play an important role in subunit communication within Fno.



Cuong Quang Le, Mercy Oyugi, Ebenezer Joseph, Toan Nguyen, Md Hasmat Ullah, Joshua Aubert, Thien Phan, Joseph Tran, and Kayunta Johnson-Winters. Biochem Biophys Rep. 2016, 9:114-120.

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^aJoseph, E.; Le, C.; Nguyen, T.; Oyugi, M.; Hossain, M.; Foss, F.; Johnson-Winters, K., Biochemistry, 2016, 55, 1082-1090.

wtFno Steady-State Kinetics Data



FO CONDITIONS 50 µM MES/NaOH (pH 6.50) buffer Temp: 22 °C FO was titrated into 0.2 µM of Fno [NADPH]=600 µM

Hossain, M. S.; Le, C. Q.; Joseph, E.; Nguyen, T. Q.; Johnson-Winters, K.; Foss, F. W., Jr. Org. Biomol. Chem, 2015, 13, 5082-5085. Joseph, E.; Le, C.; Nguyen, T.; Oyugi, M.; Hossain, M.; Foss, F.; Johnson-Winters, K. Biochemistry, 2016, 55, 1082-1090.

Enzyme	^{FO} k _{cat} (s ⁻¹)	^{ғо} К _т (µМ)	^{FO} k _{cat} /K _m (M ⁻¹ s ⁻¹)	NADPH K _{cat} (s ⁻¹)	^{NADPH} K _m (µM)	^{NADPH} k_{cat}/K_m (M ⁻¹ s ⁻¹)
<i>wt</i> Fno	5.3 ± 0.1	4.0 ± 0.4	(1.3 ± 0.3) x 10 ⁶	5.41 ± 0.04 4.16 ± 0.07	2.3 ± 0.2 61 ± 6	$(2.4 \pm 0.2) \times 10^{10}$ $(6.8 \pm 0.1) \times 10^{7}$
R186K	0.407 ± 0.008	1.3 ± 0.1	(3.1 ± 0.2) x 10 ⁵	0.35 ± 0.02	279 ± 28	(1.2 ± 0.1) x 10 ³
R186Q	0.0035 ± 0.0001	2.0 ± 0.3	(1.7 ± 0.2) x 10 ³	0.0039 ± 0.0001	139 ± 13	(2.8 ± 0.2) x 10 ¹
R186I	0.0017 ± 0.0006	4.3 ± 0.8	(3.9 ± 0.2) x 10 ²	0.0013 ± 0.0001	35 ± 8	(3.7 ± 0.2) x 10 ¹
T192V	0.0017 ± 0.0001	3.5 ± 0.4	(4.8 ± 0.6) x 10 ²	0.0010 ± 0.0003	124 ± 17	$0.8 \pm 0.2 \times 10^{1}$
T192A	0.0025 ± 0.0007	1.4 ± 0.4	(1.7 ± 0.2) x 10 ³	0.050 ± 0.001	91 ± 19	(5.4 ± 0.1) x 10 ²
S190A	0.00030 ± 0.00001	0.8 ± 0.2	(3.7 ± 0.1) x 10 ²	0.0039 ± 0.0004	49 ± 13	(7.9 ± 0.2) x 10 ¹
T09A	0.0930 ± 0.0011	0.9 ± 0.1	(1.0 ± 0.1) x 10 ⁵	0.0976 ± 0.0012	28 ± 2	(3.4 ± 0.2) x 10 ³

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	<i>К</i> _d ^{FO} (nM)	К _d ^{NADPH} (nM)	NADPH Hill Coefficient
	3.6 ± 0.7	2.0 ± 0.3	0.61 ± 0.03
	13 ± 2	64 ± 9	0.9 ± 0.1
	4.8 ± 0.5	156 ± 35	0.8 ± 0.1
	12 ± 2	53 ± 5	1.0 ± 0.1
	11 ± 2	101 ± 21	0.5 ± 0.1
	21 ± 9	156 ± 24	0.9 ± 0.1
	54 ± 15	254 ± 85	0.8 ± 0.1
	7 ± 1	138 ± 35	0.8 ± 0.1
	8 ± 3	176 ± 67	0.5 ± 0.1
_	35 ± 6	107 ± 31	0.7 ± 0.1



Fno Variant Double Reciprocal Plots





[A] Pre steady-state kinetic transfer to FO from NADPI shown for varying Fno co 1.0 µM (solid circle), 1.5 circle), and 2.0 µM (solid trian

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Binding-

- no cooperativity, diverging from the behavior of wtfno.
- pronounced cooperativity.

Steady-state-

- similar to what we see in wtfno.

Pre Steady-state-

that they are important in catalysis.

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Pre-Steady-State Kinetics Data

F		1			
	Enzyme	1 μM	1.5 μΜ	2 µM	
		<i>k</i> _{obs} (s ⁻¹)	<i>k</i> _{obs} (s⁻¹)	<i>k</i> _{obs} (s ⁻¹)	
	<i>wt</i> Fno	47.9 ± 0.5 ^a			
		1.99 ± 0.02			
	R186K	0.45 ± 0.02	0.46 ± 0.02	0.47 ± 0.02	
•	R186Q	0.27 ± 0.01	0.25 ± 0.01	0.23 ± 0.02	
- a					
~~~~	R186I	0.0040 ±	0.0050 ±	0.00680 ±	
T T		0.0006	0.00006	0.00006	
	T192A	0.2681 ±	0.2748 ±	0.2759 ± 0.0151	
		0.0046	0.0057		
0.15	T192V	0.1858 ±	0.6100 ±	0.1372 ± 0.0007	
		0.0091	0.0266		
	S190A	0.046 ±	$0.050 \pm 0.001$	$0.06 \pm 0.01$	
s of hydride		0.001			
H by Fno is	H133N	0.4116 ±	0.3856 ±	0.3518 ± 0.0083	
ncentrations:		0.0042	0.0156		
δ μΜ (open	H133A	0.0135 ±	0.0211 ±	0.0323 ±	
ngle).		0.0001	0.0001	0.0003	
	T09A	0.1709 ±	0.1735 ±	0.2083 ± 0.0040	
		0.0022	0.0010		

## Conclusions

• The Hill Coefficients for the Fno variants are around 1 for R186K, R186I, AND T192V, revealing • R186Q, S190A, and T09A variants all have a Hill coefficient of 0.8 ± 0.1, which is a less

T192A, H133A, and H133N are similar to wtfno, showing negative cooperativity

• The Fno variants R198K, R186Q, R186I, S190A, T192V and T09A did not mimic the downward and concave curvature as previously observed with *wt*Fno, but instead yielded a straight line, revealing that these Fno variants did not display cooperativity kinetics. • T192A displayed a downward and concave curvature therefore negative cooperativity which is

• The hydride transfer step for these Fno variants was significantly slower than *wt*Fno, suggesting